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## TENUIFOLIOSE Q, A NEW OLIGOSACCHARIDE ESTER FROM THE ROOT OF *POLYGALA TENUIFOLIA* WILLD.

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From the cortexes of *Polygala tenuifolia* Willd., a new oligosaccharide ester, tenuifoliose Q (1), was isolated together with three known compounds. The structure of 1 was elucidated by spectroscopic and physiochemical analysis as an oligosaccharide esterified with acetic, benzoic and *p*-hydroxycinnamoyl acid.

Keywords: Polygala tenuifolia; Oligosaccharide ester; Tenuifoliose Q

## **INTRODUCTION**

It has been reported that plenty of oligosaccharide esters have been isolated from the genus of *Polygala* [1–4]. The rarity of structures of this type of constituents inspired us to investigate further for additional oligosaccharide esters from the root of *Polygala tenuifolia* Willd., a well-known traditional Chinese medicine used as an expectorant, tonic, sedative and for preventing dementia [5]. We have already isolated four compounds with this skeleton, and one of which was a new one. Here, we report the isolation and structure elucidation of the new oligosaccharide ester tenuifoliose Q from the cortexes of *Polygala tenuifolia*, together with three known homologous compounds (Fig. 1).

## **RESULTS AND DISCUSSION**

The n-BuOH-soluble parts of the 95% EtOH extract of *P. tenuifolia* were subjected to macroporous resin D101 column chromatography, eluted with gradient EtOH $-H_2O$ . The 50% aq. EtOH eluate was chromatographed on silica gel, ODS column and by HPLC to afford tenuifoliose Q, along with three known oligosaccharide esters.

Tenuifoliose Q (1) was obtained as a white amorphous powder and its molecular formula deduced to be  $C_{65}H_{82}O_{37}$  from its TOF-MS (1477 [M + Na]<sup>+</sup>) and by <sup>13</sup>C NMR analysis.

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FIGURE 1 Structures of compounds 1-4.

The IR spectrum of 1 showed the presence of hydroxyl groups  $(3405 \text{ cm}^{-1})$ , carbonyl group  $(1720 \text{ cm}^{-1})$ , double bond  $(1633 \text{ cm}^{-1})$ , and aromatic rings  $(1604, 1513, 1452 \text{ cm}^{-1})$ . The <sup>1</sup>H NMR spectrum of **1** (see Table I) showed a group of benzoyl proton signals at  $\delta$  8.21 (2H, dd, J = 8.0, 1.5 Hz), 7.60 (2H, t, J = 8.0 Hz) and 7.70 (1H, tt, J = 8.0, 1.0 Hz); two groups of p-hydroxycinnamoyl proton signals at  $\delta$  7.37 (2H, d, J = 8.0 Hz), 6.79 (2H, d, J = 8.5 Hz), 6.33 (1H, d, J = 16.0 Hz), 7.64 (1H, d, J = 16.5 Hz), 7.13 (2H, d, J = 8.5 Hz), 7.54 (2H, d, J = 8.5 Hz), 6.32 (1H, d, J = 15.5 Hz) and 7.60 (1H, d, J = 15.5 Hz), together with two acetyl proton signals at  $\delta$  1.98 and 1.51 (3H each, s). In addition, in the <sup>1</sup>H NMR spectrum of 1, there are five anomeric proton signals at  $\delta$  5.86 (1H, d, J = 3.0 Hz), 4.57 (1H, d, J = 7.5 Hz), 4.50 (1H, d, J = 8.0 Hz) 4.45 (1H, d, J = 7.5 Hz), 5.53 (1H, brs) and a H-3 proton signal of fructose at  $\delta$  5.71 (1H, d, J = 8.5 Hz). All these data suggested that 1 is an oligosaccharide esterified with acetic, benzoic and *p*-hydroxycinnamoyl acids. On acid hydrolysis, 1 gave glucose and fructose. The NMR data of 1 are very similar to those of tenuifoliose L [1], except that an acetyl signals at  $\delta$  2.06 of tenuifoliose L disappeared in 1, and the C-6 signal of Glc-1 in 1 shifted 1.7 ppm upfield, while the C-5 signal shifted 3.3 ppm downfield. Therefore, we deduced that the C-6 position of Glc-1 in 1 was perhaps deacetylated. All proton signals were assigned through the TOCSY experiment, and we found that the H-6 signals of Glc-1 in 1 at  $\delta$  3.57 (1H, dd, J = 11.5, 5.0 Hz) and  $\delta$  3.69 (1H, dd, J = 11.5, 2.0 Hz) shifted 0.58 and 0.51 ppm upfield, compared with those signals in tenuifoliose L. Moreover, all the proton signals of Glc-1 in 1 were the same as those of tenuifoliose J [1], which does not have an acetyl group at H-6 of Glc-1. Thus, the structure of 1 was concluded to be that shown in Fig. 1.

Compounds 2-4 were identified by comparing their physical and spectral data (Table II) with the literature values, as tenuifoliose L (2), tenuifoliose I (3) and tenuifoliose H (4) [1].

## **EXPERIMENTAL**

#### **General Experimental Procedures**

Optical rotations were measured on a Polartronic D polarimeter. UV spectra were recorded on a TU-1901 spectrophotometer, whereas IR spectra were obtained on an AVATER-360 spectrophotometer. MDLDI-TOF MS spectra were performed with a LDI 1700 spectrometer, using CCA ( $\alpha$ -cyano-4-hydroxycinnamic acid) as matrix, while ESI MS spectra were performed on a QSTAR mass spectrometer. <sup>1</sup>H, <sup>13</sup>C NMR, TOCSY, HMQC and HMBC spectra were measured on a Bruker AM-500 spectrometer. D101 resin was from Tianjin

	I	2	3	4
	5.86 (1H, d, <i>J</i> = 3.0 Hz)	5.89 (IH, d, <i>J</i> = 3.5 Hz)	6.62 (1H, brs)	6.62 (1H, d, $J = 3.5$ Hz)
	4.57 (1H, d, $J = 7.5$ Hz)	4.58 (1H, d, $J = 7.5$ Hz)	5.15 (1H, d, $J = 7.0$ Hz)	5.10 (1H, d, $J = 8.0 \mathrm{Hz}$ )
	4.50 (1H, d, $J = 8.0 \mathrm{Hz}$ )	4.52 (1H, d, $J = 8.0 \mathrm{Hz}$ )	5.26 (1H, d, J = 8.0 Hz)	5.23 (1H, d, $J = 8.0  \text{Hz}$ )
	4.45 (1H, d, <i>J</i> = 7.5 Hz)	4.45 (1H, d, $J = 8.0 \mathrm{Hz}$ )	5.08 (1H, d, J = 7.5 Hz)	5.09 (1H,d, $J = 7.5$ Hz)
	4.26 (1H, d, $J = 12.0$ Hz) 4.63 (1H, d, $J = 12.0$ Hz) 5.71 (1H, d, $J = 8.5$ Hz)	4.27 (1H, d, $J = 12.0$ Hz) 4.60 (1H, d, $J = 12.0$ Hz) 5.72 (1H, d, $J = 8.0$ Hz)	$\begin{array}{l} 4.81 \ (1H,  d,  J = 10.5  \text{Hz}) \\ 5.41 \ (1H,  d,  J = 12.0  \text{Hz}) \\ 6.53 \ (1H,  d,  J = 8.0  \text{Hz}) \end{array}$	4.85 (1H, d, <i>J</i> = 12.0Hz 5.39 (1H, d, <i>J</i> = 12.0Hz 6.53 (1H, d, <i>J</i> = 8.0Hz)
	5.53 (1H, brs) 1.23 (3H, d, <i>J</i> = 6.0 Hz) 1.98 (3H, s) 1.51 (3H, s)	5.53 (1H, d, <i>J</i> = 1.5 Hz) 1.22 (3H, d, <i>J</i> = 6.0 Hz) 2.06 (3H, s) 1.98 (3H, s) 1.51 (3H, s)	2.14 (3H, s) 1.63 (3H, s)	2.15 (3H, s) 2.19 (3H, s) 1.76 (3H, s)
3)	8.21 (2H, dd, $J = 8.0, 1.5$ Hz) 7.60 (2H, t, $J = 8.0$ Hz) 7.70 (1H, tt, $J = 8.0, 1.0$ Hz)	8.19 (2H, dd, $J = 8.0$ , 1.5 Hz) 7.60 (2H, t, $J = 8.0$ Hz) 7.70 (1H, tt, $J = 8.0$ , 1.0 Hz)	8.35 (2H, d, <i>J</i> = 6.5 Hz) 7.56 (2H, m) 7.58 (1H, m)	8.37 (2H, d, $J = 7.0$ Hz) 7.58 (2H, d, $J = 7.0$ Hz) 7.60 (1H, t, $J = 6.5$ Hz)
K2)	7.37 (2H, d, $J = 8.0 \text{ Hz}$ ) 6.79 (2H, d, $J = 8.5 \text{ Hz}$ ) 6.33 (1H, d, $J = 16.0 \text{ Hz}$ ) 7.64 (1H, d, $J = 16.5 \text{ Hz}$ )	7.35 (2H, d, $J = 8.5$ Hz) 6.79 (2H, d, $J = 9.0$ Hz) 6.31 (1H, d, $J = 16.0$ Hz) 7.63 (1H, d, $J = 16.0$ Hz)	7.39 (2H, d, $J = 7.5$ Hz) 7.14 (2H, d, $J = 7.5$ Hz) 6.66 (1H, d, $J = 16.0$ Hz) 7.92 (1H, d, $J = 16.0$ Hz)	7.40 (2H, d, $J = 8.5$ Hz) 7.14 (2H, d, $J = 8.0$ Hz) 6.66 (1H, d, $J = 16.0$ Hz 7.96 (1H, d, $J = 16.0$ Hz
R4)	7.54 (2H, d, $J = 8.5$ Hz) 7.13 (2H, d, $J = 8.5$ Hz) 6.32 (1H, d, $J = 15.5$ Hz) 7.60 (1H, $3 - 15.5$ Hz)	7.54 (2H, d, $J = 9.0 \text{ Hz}$ ) 7.13 (2H, d, $J = 8.5 \text{ Hz}$ ) 6.30 (1H, d, $J = 16.0 \text{ Hz}$ )	7.65 (2H, d, $J = 7.5$ Hz) 7.21 (2H, d, $J = 7.5$ Hz) 6.45 (1H, d, $J = 16.0$ Hz) 7.20 (1H, d, $J = 16.0$ Hz)	7.67 (2H, d, $J = 8.5$ Hz) 7.22 (2H, d, $J = 8.5$ Hz) 6.47 (1H, d, $J = 16.0$ Hz 6.47 (1H, d, $J = 16.0$ Hz

TABLE I  $^{-1}$ H NMR data of compounds 1-4 (500 MHz)<sup>a</sup>

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		1	2	3	4	Tenuifoliose L <sup>[4]</sup>
Glc-1	1	93.5	92.7	91.9	91.8	92.7
	2	81.9	81.3	81.4	81.1	81.3
	3	80.0	79.3	78.4	78.3	79.3
	4	70.8	70.5	69.7	69.8	70.5
	5	72.9	69.6	69.0	68.8	69.6
	6	62.6	64.3	63.5	63.3	64.3
Glc-2	1	105.9	105.4	105.6	106.2	105.3
	2	75.3	75.3	75.4	75.4	75.8
	3	79.0	78.5	78.3	78.9	78.5
	4	72.1	71.5	71.3	71.6	71.5
	5	78.3	77.7	77.9	78.0	77.7
	6	63.6	63.1	63.0	63.0	63.1
Glc-3	1	104.7	104.1	104.2	103.8	104.1
	2	75.9	75.4	74.1	73.3	75.4
	3	83.9	83.4	88.0	84.1	83.4
	4	70.1	69.6	69.0	69.1	69.6
	5	72.9	72.4	73.9	71.8	72.4
	6	63.6	63.1	63.6	63.0	63.1
Glc-4	1	105.9	105.4	105.4	105.7	105.3
	2	76.3	75.8	75.4	76.0	75.7
	3	78.9	78.4	77.9	78.0	78.4
	4	72.1	71.5	71.3	71.3	71.5
	5	78.4	77.9	78.4	78.0	77.9
-	6	63.4	62.8	62.2	62.4	62.8
Fru	1	66.8	66.3	64.9	65.1	66.3
	2	104.2	103.7	103.5	103.4	103.7
	3	80.9	80.4	79.6	79.8	80.4
	4	/4.2	74.0	/3.9	/3.3	74.0
	5	85.1	84.7	84.8	84.9	84.7
Dham	0	03.8	03.9	03.3	02.0	03.9
Knam	1	99.5	99.0 71.0			100.1
	2	72.4	71.9			/1.8
	5	74.1	12.2			72.1
	4	74.1	75.7			75.7
	5	19.6	10.9			10.9
$\Lambda_{C}(\mathbf{P5})$	0	18.0	172.5	170.7	170.7	172.5
AC(R3)	2		20.8	20.7	20.8	20.8
$A_{c}(\mathbf{R6})$	1	172.5	172.0	20.7	170.1	172.0
AC(RO)	2	21.6	21.1		20.9	21.1
Ac(R7)	- 1	172.7	172.2	170.7	170.6	172.2
ne(ner)	2	20.9	20.4	20.4	20.4	20.4
$B_{Z}(R_{3})$	1	131.5	131.0	130.3	130.3	131.0
()	2.6	131.6	131.0	130.3	130.3	131.0
	3.5	130.5	130.0	129.0	129.1	130.0
	4	135.4	134.9	133.8	133.8	134.9
	α	167.7	167.1	166.3	166.3	167.1
Cinn(R2)	1	127.6	127.1	125.9	125.8	127.1
	2,6	131.8	131.2	130.7	130.8	131.2
	3,5	117.5	116.9	116.6	116.9	116.9
	4	161.9	161.4	161.5	161.7	161.3
	α	168.9	168.4	166.8	166.8	168.3
	β	115.4	114.9	114.6	114.6	114.9
	γ	147.4	146.9	145.8	145.6	146.9
Cinn(R4)	1	130.1	129.6	125.8	125.8	129.5
	2,6	131.7	131.1	130.7	130.8	131.1
	3,5	118.4	117.9	116.8	116.7	117.9
	4	160.2	159.7	161.6	161.5	159.7
	α	168.2	167.5	166.6	166.5	167.5
	β	117.4	116.7	114.5	114.6	116.7
	γ	146.4	146.1	145.6	145.8	146.1

TABLE II  ${}^{13}$ C NMR data of compounds  $1-4 (125 \text{ MHz})^{a}$ 

 $^{a}$  Compounds 1 and 2 were measured in MeOH, while compounds 3 and 4 were measured in  $\mathrm{C}_{5}\mathrm{D}_{5}\mathrm{N}$ . and the signal assignments of 3 and 4 were aided by COSY, HMQC and HMBC spectra.

Chemical Co., and the column chromatography silica gel (200–300 mesh) was supplied by the Qingdao Marine Chemical Factory.

## **Plant Material**

The cortexes of *P. tenuifolia* were collected from Shanxi Province, in October 2000. The plant was identified by one of the authors (P.F. Tu). A voucher specimen (No. 001020) is deposited in the herbarium of School of Pharmaceutical Sciences, Peking University, Beijing, China.

#### **Extraction and Isolation**

The air-dried roots of *P. tenuifolia* (11 kg) were ground and refluxed three times with 95% EtOH (77 L). The 95% EtOH solution was then combined and evaporated *in vacuo* to yield 4.9 kg of residue, a portion (2 kg) of which was suspended in water and extracted successively with light petroleum, CHCl<sub>3</sub> and n-BuOH. Parts of the n-BuOH extract (325 g) were subjected to a macroporous resin D101 column (11.5 × 85.5 cm). The adsorbed material was eluted with H<sub>2</sub>O, 20, 50, 70, and 95% EtOH, respectively. The 50% EtOH eluate (78 g) was chromatographed on silica gel (1.6 kg), eluting with CHCl<sub>3</sub>—MeOH—H<sub>2</sub>O in a gradient manner (500 : 1 : 0 → 60 : 40 : 5). Fractions 79–95 (4 g) was first subjected to ODS column chromatography, then purified by HPLC with MeOH—H<sub>2</sub>O (50:50) as mobile phase to furnish **3** (45.6 mg) and **4** (139.2 mg). Fractions 96—114 (3 g) were also first subjected to ODS column chromatography, MeOH—H<sub>2</sub>O (20 : 80 → 80 : 20) as the eluting solvent, to give 20 fractions; fractions 9–10 were then further subjected to Sephadex LH-20 and finally purified by HPLC, with MeOH—H<sub>2</sub>O (50:50) as mobile phase to furnish **1** (46.9 mg).

Tenuifoliose Q (1), white amorphous  $[\alpha]_D^{24}$  powder, -92.8 (c = 0.85, MeOH). IR  $\nu_{max}^{KBr}$  (cm<sup>-1</sup>): 3405 (OH), 1720 (C=O), 1633 (C=C), 1604, 1513, 1452 (aromatic ring). TOF-MS (CCA as matrix) m/z: 1477 [M + Na]<sup>+</sup>. <sup>1</sup>H NMR data: Table I. <sup>13</sup>C NMR data: Table II.

Tenuifoliose L (2), white amorphous powder,  $[\alpha]_D^{24} - 46.4$  (c = 0.85, MeOH). TOF-MS (CCA as matrix) m/z: 1519 [M + Na]<sup>+</sup>. <sup>1</sup>H NMR data: Table I. <sup>13</sup>C NMR data: Table II.

Tenuifoliose I (3), white amorphous powder,  $[\alpha]_D^{24} - 10.7$  (c = 0.89, MeOH). ESI-MS m/z: 1369 [M + 18]<sup>+</sup>. <sup>1</sup>H NMR data: Table I. <sup>13</sup>C NMR data: Table II.

Tenuifoliose H (4), white amorphous powder,  $[\alpha]_D^{24} - 60.2$  (c = 0.82, MeOH). ESI-MS m/z: 1327[M + 18]<sup>+</sup>. <sup>1</sup>H NMR data: Table I. <sup>13</sup>C NMR data: Table II.

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